Chemistry 11 Lab 2A: Making Observation	Name: Date: Block:
For Students:	For Teacher:

For Students:	For Teacher:			
Lab performed:	Pre-lab completion:	Yes	No	
Lab due:	Lab Submitted:	On Time	Late	

Lab performed:	Pre-lab completion:	Yes	No
Lab due:	Lab Submitted:	On Time	Late
Introduction & Objectives			J
A qualitative observation is:			
A quantitative observation is:			
Objectives:			
2.			
Answer the following questions:			
 What are some examples of qualitative specific examples 	ve observations that you could make	during this lab	? Provide at least 2
What are some examples of quantitat	tive observations that you could mal	ze during this la	ah? Provide at least
2 specific examples	in a contained state you could make		

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Part I: Zinc Metal with Water Procedure

Table I: Zinc Metal with Water

	Qualitative Observations	Quantitative Observations
Before		
During		
After		

Part II: Calcium Metal with Water Procedure

Table II: Calcium Metal with Water

	Qualitative Observations	Quantitative Observations
Before		
During		
After		

Part III: Aluminum metal with Copper (II) chloride Procedure

Table III: Aluminum metal with Copper (II) chloride

	Qualitative Observations	Quantitative Observations
Before		
During		
After		

Analysis of Results

Answer	the	follo	wina	in	full	sentences

1. In which of the tests did you observe a colour change?
2. a. In which of the tests did you observe the formation of a gas?
b. What observations allowed you to conclude that a gas was produced?
3. In which test(s) did you observe no change taking place?
Conclusion
State the results of Objectives 1 and 2
1.
2.